



Seat No. \_\_\_\_\_

**HJ-003-1014004**  
**B. Sc. (Sem.-IV) (CBCS)**  
**(W.E.F. 2016) Examination**  
**April - 2023**  
**C-401 : Chemistry Theory**  
*(Old Course)*

**Faculty Code : 003**

**Subject Code : 1014004**

Time :  $2\frac{1}{2}$  Hours / Total Marks : 70

**Instructions :**

- (1) This question paper contains five questions.
- (2) All are compulsory.
- (3) Figures to the right indicate full marks.

- 1 (a) Answer the following questions : 4
- (i) Define : Organometallic compounds.
  - (ii) Ferrocene has \_\_\_\_\_ type structure.
  - (iii) Give structure of porphyrin.
  - (iv) In Hemoglobin iron has \_\_\_\_\_ oxidation state.
- (b) Answer any one : 2
- (1) Give preparation method of organoaluminium compounds.
  - (2) How lead poisoning is cured ?
- (c) Answer any one : 3
- (1) Short note on "Zeise salt".
  - (2) Explain myoglobin.
- (d) Answer any one : 5
- (1) Explain the structure of trimethyl aluminium (dimer).
  - (2) Describe the structure and role of chlorophyll.



- 4 (a) Answer the following questions : 4
- (1) Give structure of Wittig reagent.
  - (2) Give structure of Benzillic acid.
  - (3) Define Molar volume.
  - (4) Give the name of method to determine surface tension.
- (b) Answer any one : 2
- (1) Explain principle of Hofmann degradation reaction.
  - (2) Write short note on Parachor.
- (c) Answer any one : 3
- (1) Give mechanism of Perkin reaction.
  - (2) Parachor of ethane and propane are 110.5 and 150.8 respectively. Calculate parachor of Hexane.
- (d) Answer any one : 5
- (1) Explain Aldol condensation reaction with principle, mechanism and application.
  - (2) Define refractive index and give details on Abbe refractometer.
- 5 (a) Answer the following questions : 4
- (1) Define : Open system.
  - (2) Define Specific heat with equation.
  - (3) Define : Extensive properties.
  - (4) Give name of the process in which pressure remains constant.
- (b) Answer any one : 2
- (1) State the first law of thermodynamics obtain the mathematical expression.
  - (2) Derive the relationship :  

$$\Delta H = \Delta U + \Delta nRT$$
- (c) Answer any one : 3
- (1) Write note on applications and limitations of thermodynamics.
  - (2) Distinguish between Reversible and Irreversible process.
- (d) Answer any one : 5
- (1) Explain  $C_p$  and  $C_v$ ; state their relation.
  - (2) Describe different types of thermodynamics processes.